

(2.11) A gas centrifuge plant is set up in order to enrich UF_6 of natural isotopic composition in ^{235}U . The centrifuges, which each have a length of 100 cm and a diameter of 20 cm, rotate at 40000 rpm. The gas temperature is 70 deg.C.

(a) Prove that the separation factor α in (2.47) can be approximated by e^δ according to (2.58) when the product flow is very small compared to the waste flow, and α is not far from 1.

$e^\delta = 1 + \delta + \delta^2/2 + \dots$ But $\alpha = 1 + \delta$. Neglecting higher order terms we get $e^\delta = \alpha$. Q.E.D.

(b) Using this approximation, what is the theoretical separation factor for one unit?

$$u := 1.660540 \cdot 10^{-27} \cdot \text{kg} \quad R := 0.08206 \cdot \text{liter} \cdot \text{atm} \cdot \text{mole}^{-1} \cdot \text{K}^{-1} \quad T := (70 + 273.15) \cdot \text{K}$$

$$M_H := (238 + 6 \cdot 19) \cdot \frac{\text{gm}}{\text{mole}} \quad M_L := (235 + 6 \cdot 19) \cdot \frac{\text{gm}}{\text{mole}}$$

$$\omega := 2 \cdot \pi \cdot \frac{40000}{60} \cdot \text{sec}^{-1} \quad r := \frac{20}{2} \cdot 0.01 \cdot \text{m} \quad v_{\text{rot}} := \omega \cdot r \quad v_{\text{rot}} = 418.879 \cdot \text{m} \cdot \text{sec}^{-1}$$

$$\delta := (M_H - M_L) \cdot \omega^2 \cdot \frac{r^2}{R \cdot T \cdot 2} \quad \text{OBS! } 1/1000 \text{ omitted because of automatic unit conversion used.}$$

$$\delta = 0.092 \quad \alpha := \exp(\delta) \quad \alpha = 1.097$$

(c) Assuming that the enrichment factor obtained with the centrifuge is only 70% of the theoretical one, what number of units would be required in series in order to achieve UF_6 with 3% ^{235}U ?

$$\alpha_{\text{eff}} := \frac{70}{100} \cdot (\alpha - 1) + 1 \quad x_p := \frac{3}{100} \quad x_f := \frac{0.724}{100} \quad \alpha_{\text{eff}} = 1.068$$

$$Np := \frac{\ln \left[\frac{x_p \cdot (1 - x_f)}{x_f \cdot (1 - x_p)} \right]}{\ln(\alpha_{\text{eff}})} \quad Np = 22.073 \quad \text{but } Np \text{ must be integer, hence: } Np = 22$$