

(6.10) An experiment is done with ^{60}mCo which emits 0.05860 MeV γ . The detector used is a NaI crystal. What photo peaks will be observed if the electron binding energies in sodium are K 1072 and L 63 eV, and in iodine K 33170 and L 4800 eV?

$$eV := 1.60217733 \cdot 10^{-19} \cdot \text{joule} \quad keV := 1000 \cdot eV \quad MeV := 1000 \cdot keV$$

$$E_{\gamma} := 0.05860 \cdot MeV \quad \text{This is the full } \gamma\text{-energy}$$

The observed lines correspond to the photo electron energies. Hence:

$$E_b := 1072 \cdot eV \quad E_e := E_{\gamma} - E_b \quad \text{Eqn. (6.23)} \quad E_e = 57.53 \cdot keV \quad \text{Na}_K$$

$$E_b := 63 \cdot eV \quad E_e := E_{\gamma} - E_b \quad E_e = 58.54 \cdot keV \quad \text{Na}_L$$

$$E_b := 33170 \cdot eV \quad E_e := E_{\gamma} - E_b \quad E_e = 25.43 \cdot keV \quad \text{I}_K$$

$$E_b := 4800 \cdot eV \quad E_e := E_{\gamma} - E_b \quad E_e = 53.80 \cdot keV \quad \text{I}_L$$