

(8.6) In a GM counter, sample A gave 12630, B 15480, and A+B together 25147 cpm. (a) What is the resolving time of the counter? With the same counter, the distribution of radioactive samarium between an organic phase and water was measured according to $D_m = R_{org}/R_{aq}$. The measured R_{org} is 37160 cpm, and that of R_{aq} is 2965. (b) What is the measured D_m ? (c) Using the corrections for the resolving time, what is the true D -value?

First some definitions of useful units:

$$\mu s := 10^{-6} \cdot \text{sec} \quad \text{cpm} := \text{min}^{-1}$$

Ignore the background count rate in calculating the dead-time. $R_0 := 0 \cdot \text{cpm}$

$$R_A := 12630 \cdot \text{cpm} \quad R_B := 15480 \cdot \text{cpm} \quad R_{AB} := 25147 \cdot \text{cpm}$$

$$x := R_A \cdot R_B - R_{AB} \cdot R_0 \quad y := R_A \cdot R_B \cdot (R_{AB} + R_0) - R_{AB} \cdot R_0 \cdot (R_A + R_B)$$

$$z := y \cdot \frac{R_A + R_B - R_{AB} - R_0}{x^2} \quad t_r := x \cdot \frac{1 - \sqrt{1 - z}}{y}$$

$$(a) \quad t_r = 5.089 \cdot 10^{-4} \cdot \text{sec} \quad t_r = 509 \cdot \mu s$$

$$(b) \quad R_{org} := 37160 \cdot \text{cpm} \quad R_{aq} := 2965 \cdot \text{cpm}$$

$$D := \frac{R_{org}}{R_{aq}} \quad D = 12.53$$

$$(c) \quad R_{orgcorr} := \frac{R_{org}}{1 - R_{org} \cdot t_r} \quad R_{aqcorr} := \frac{R_{aq}}{(1 - R_{aq} \cdot t_r)}$$

$$D_{corr} := \frac{R_{orgcorr}}{R_{aqcorr}} \quad D_{corr} = 17.84$$