

(14.9) A 2 MeV neutron collides elastically with an iron atom (^{56}Fe). What is the average temperature (corresponding to the maximum velocity) which can be ascribed to the iron nucleus after the collision?

Constants, known values, and units:

$$\text{MeV} := 1.60217733 \cdot 10^{-13} \cdot \text{joule} \quad \text{amu} := 1.6605402 \cdot 10^{-27} \cdot \text{kg} \quad M_n := 1 \cdot \text{amu}$$

$$M_{\text{Fe}} := 56 \cdot \text{amu} \quad k := 1.380658 \cdot 10^{-23} \frac{\text{joule}}{\text{K}}$$

Given in the text:

$$E_n := 2 \cdot \text{MeV}$$

Calculations:

$$V_n := \left(\frac{E_n}{\frac{1}{2} \cdot M_n} \right)^{\frac{1}{2}} \quad V_n = 1.965 \cdot 10^7 \cdot \text{m} \cdot \text{sec}^{-1}$$

Moment law:

$$v_n \cdot M_n = v_n \cdot M_n + v_{\text{Fe}} \cdot M_{\text{Fe}} \quad v_n = V_n - \frac{v_{\text{Fe}} \cdot M_{\text{Fe}}}{M_n}$$

Energy law:

$$E_n = \frac{1}{2} \cdot M_n \cdot v_n^2 + \frac{1}{2} \cdot M_{\text{Fe}} \cdot v_{\text{Fe}}^2 \quad E_n = \frac{1}{2} \cdot M_n \cdot \left(V_n - \frac{v_{\text{Fe}} \cdot M_{\text{Fe}}}{M_n} \right)^2 + \frac{1}{2} \cdot M_{\text{Fe}} \cdot v_{\text{Fe}}^2$$

$$v_{\text{Fe}} := \frac{1}{2 \cdot \left(\frac{-1 \cdot M_{\text{Fe}}^2}{2 \cdot M_n} - \frac{1}{2} \cdot M_{\text{Fe}} \right)} \cdot \left(-V_n \cdot M_{\text{Fe}} - \sqrt{M_{\text{Fe}} \cdot \frac{\sqrt{2 \cdot M_{\text{Fe}} \cdot E_n + 2 \cdot M_n \cdot E_n - M_n^2 \cdot V_n^2}}{\sqrt{M_n}}} \right)$$

$$v_{\text{Fe}} := \frac{2 \cdot \sqrt{2 \cdot M_n \cdot E_n}}{M_n + M_{\text{Fe}}} \quad v_{\text{Fe}} = 6.893 \cdot 10^5 \cdot \text{m} \cdot \text{sec}^{-1}$$

$$E_{\text{Fe}} := \frac{1}{2} \cdot M_{\text{Fe}} \cdot v_{\text{Fe}}^2 \quad \text{This is non-relativistic}$$

$$E_{\text{Fe}} := 4 \cdot \frac{M_{\text{Fe}} \cdot M_n}{(M_{\text{Fe}} + M_n)^2} \cdot E_n \quad E_{\text{Fe}} = 0.138 \cdot \text{MeV}$$

$$\text{Use: } E = 3kT/2, \text{ eqn. (2.21):} \quad T_{\text{Fe}} := \frac{2 \cdot E_{\text{Fe}}}{3 \cdot k} \quad T_{\text{Fe}} = 1.1 \cdot 10^9 \cdot \text{K}$$