

(18.5) Ten mg ^{238}U has been collected in the kidneys. Considering the biological half-life of uranium and assuming only one α -emission in ^{238}U decay, calculate the dose (in Sv) received by the organs if the uranium is evenly distributed. The weight of a kidney is 150 g.

Constants, data, and units:

$$\begin{array}{lll}
 N_A := 6.022137 \cdot 10^{23} \cdot \text{mole}^{-1} & M_U := 238 \cdot \text{gm} \cdot \text{mole}^{-1} & \text{MeV} := 1.6021773 \cdot 10^{-13} \cdot \text{joule} \\
 \text{Gy} := \text{joule} \cdot \text{kg}^{-1} & \text{Sv} := \text{joule} \cdot \text{kg}^{-1} & \text{mSv} := 0.001 \cdot \text{Sv} \\
 w_{R\alpha} := 20 & t_{\text{biol}} := 15 \cdot \text{day} & E_{\alpha} := 4.198 \cdot \text{MeV}
 \end{array}$$

Data from the text:

$$m_{\text{kidney}} := 2 \cdot 150 \cdot \text{gm} \quad 2 \text{ kidneys} \quad m_U := 10 \cdot \text{mg}$$

Calculations:

$$\lambda_{\text{biol}} := \frac{\ln(2)}{t_{\text{biol}}} \quad t_{\text{mean}} := \frac{1}{\lambda_{\text{biol}}}$$

$$t_{\text{half}} := 4.468 \cdot 10^9 \cdot \text{yr} \quad \lambda_{\text{phys}} := \frac{\ln(2)}{t_{\text{half}}}$$

$$N_{0U} := \frac{m_U}{M_U} \cdot N_A \quad R_{0U} := \lambda_{\text{phys}} \cdot N_{0U} \quad R_{0U} = 124.392 \cdot \text{sec}^{-1}$$

$$D_{\text{kidney}} := \frac{t_{\text{mean}} \cdot R_{0U} \cdot E_{\alpha}}{m_{\text{kidney}}} \quad D_{\text{kidney}} = 5.214 \cdot 10^{-4} \cdot \text{Gy}$$

$$H_{\text{kidney}} := D_{\text{kidney}} \cdot w_{R\alpha} \quad H_{\text{kidney}} = 10 \cdot \text{mSv}$$